

**Reference solution (a).** Dissolve 5 mg of the substance to be examined and 10 µl of 2-pyrrolidone R in a mixture of 10 volumes of acetonitrile R1 and 90 volumes of water R and dilute to 100.0 ml with the same mixture of solvents.

**Reference solution (b).** Dilute 1.0 ml of test solution (a) to 100.0 ml with a mixture of 10 volumes of acetonitrile R1 and 90 volumes of water R. Dilute 5.0 ml of this solution to 50.0 ml with a mixture of 10 volumes of acetonitrile R1 and 90 volumes of water R.

**Reference solution (c).** Dissolve 50.0 mg of piracetam CRS in a mixture of 10 volumes of acetonitrile R1 and 90 volumes of water R and dilute to 100.0 ml with the same mixture of solvents. Dilute 10.0 ml of this solution to 50.0 ml with a mixture of 10 volumes of acetonitrile R1 and 90 volumes of water R.

**Column:**

- size:  $l = 0.25$  m,  $\varnothing = 4.6$  mm,
- stationary phase: end-capped octadecylsilyl silica gel for chromatography R (5 µm).

**Mobile phase:** mix 10 volumes of acetonitrile R1 and 90 volumes of a 1.0 g/l solution of dipotassium hydrogen phosphate R; adjust to pH 6.0 with dilute phosphoric acid R.

**Flow rate:** 1.0 ml/min.

**Detection:** spectrophotometer at 205 nm.

**Injection:** 20 µl of test solution (a) and reference solutions (a) and (b).

**Run time:** 8 times the retention time of piracetam.

**Relative retention** with reference to piracetam (retention time = about 4 min): impurity D = about 0.8; impurity A = about 1.15; impurity B = about 2.8; impurity C = about 6.3.

**System suitability:** reference solution (a):

- resolution: minimum 3.0 between the peaks due to piracetam and impurity A,
- symmetry factor: maximum 2.0 for the peak due to piracetam.

**Limits:**

- impurities A, B, C, D: for each impurity, not more than the area of the principal peak in the chromatogram obtained with reference solution (b) (0.1 per cent),
- unspecified impurities: for each impurity, not more than the area of the principal peak in the chromatogram obtained with reference solution (b) (0.1 per cent),
- total: not more than 3 times the area of the principal peak in the chromatogram obtained with reference solution (b) (0.3 per cent),
- disregard limit: 0.5 times the area of the principal peak in the chromatogram obtained with reference solution (b) (0.05 per cent).

**Heavy metals (2.4.8):** maximum 10 ppm.

Dissolve 2.0 g in 20 ml of water R. 12 ml of the solution complies with test A. Prepare the reference solution using lead standard solution (1 ppm Pb) R.

**Loss on drying (2.2.32):** maximum 1.0 per cent, determined on 1.000 g by drying in an oven at 100-105 °C.

**Sulphated ash (2.4.14):** maximum 0.1 per cent, determined on 1.0 g.

**ASSAY**

Liquid chromatography (2.2.29) as described in the test for related substances with the following modification.

**Injection:** test solution (b) and reference solution (c).

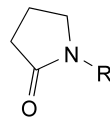
Calculate the percentage content of  $C_6H_{10}N_2O_2$  from the areas of the peaks and the declared content of piracetam CRS.

**STORAGE**

Protected from light.

**IMPURITIES**

**Specified impurities:** A, B, C, D.



A. R = H: pyrrolidin-2-one (2-pyrrolidone),

B. R =  $CH_2COOCH_3$ : methyl (2-oxopyrrolidin-1-yl)acetate,

C. R =  $CH_2COOC_2H_5$ : ethyl (2-oxopyrrolidin-1-yl)acetate,

D. R =  $CH_2CO_2H$ : (2-oxopyrrolidin-1-yl)acetic acid.

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## POLYSORBATE 80

### Polysorbatum 80

**DEFINITION**

Mixture of partial esters of fatty acids, mainly *Oleic acid (0799)*, with sorbitol and its anhydrides ethoxylated with approximately 20 moles of ethylene oxide for each mole of sorbitol and sorbitol anhydrides.

**CHARACTERS**

**Appearance:** oily, yellowish or brownish-yellow, clear or slightly opalescent liquid.

**Solubility:** dispersible in water, in anhydrous ethanol, in ethyl acetate and in methanol, practically insoluble in fatty oils and in liquid paraffin.

**Relative density:** about 1.10.

**Viscosity:** about 400 mPas at 25 °C.

**IDENTIFICATION**

**First identification:** A, D.

**Second identification:** B, C, D, E.

A. Infrared absorption spectrophotometry (2.2.24).

**Comparison:** Ph. Eur. reference spectrum of polysorbate 80.

B. Hydroxyl value (see Tests).

C. Saponification value (see Tests).

D. Composition of fatty acids (see Tests).

E. Dissolve 0.1 g in 5 ml of methylene chloride R. Add 0.1 g of potassium thiocyanate R and 0.1 g of cobalt nitrate R. Stir with a glass rod. The solution becomes blue.

**TESTS**

**Acid value (2.5.1):** maximum 2.0.

Dissolve 5.0 g in 50 ml of the prescribed mixture of solvents.

**Hydroxyl value (2.5.3, Method A):** 65 to 80.

**Peroxide value:** maximum 10.0.

Introduce 10.0 g into a 100 ml beaker, dissolve with glacial acetic acid R and dilute to 20 ml with the same solvent. Add 1 ml of saturated potassium iodide solution R and allow to stand for 1 min. Add 50 ml of carbon dioxide-free water R and a magnetic stirring bar. Titrate with 0.01 M sodium

*thiosulphate*, determining the end-point potentiometrically (2.2.20). Carry out a blank titration.

Determine the peroxide value using the following expression:

$$\frac{(n_1 - n_2) \times M \times 1000}{m}$$

- $n_1$  = volume of 0.01 M sodium thiosulphate required for the substance to be examined, in millilitres,  
 $n_2$  = volume of 0.01 M sodium thiosulphate required for the blank, in millilitres,  
 $M$  = molarity of the sodium thiosulphate solution, in moles per litre,  
 $m$  = mass of substance to be examined, in grams.

**Saponification value** (2.5.6): 45 to 55, determined on 4.0 g.

Use 30.0 ml of 0.5 M alcoholic potassium hydroxide, heat under reflux for 60 min and add 50 ml of anhydrous ethanol R before carrying out the titration.

**Composition of fatty acids.** Gas chromatography (2.4.22, Method C). Use the mixture of calibrating substances in Table 2.4.22-3.

**Column:**

- material: fused silica,
- size:  $l = 30$  m,  $\varnothing = 0.32$  mm,
- stationary phase: macrogol 20 000 R (film thickness 0.5  $\mu$ m).

Carrier gas: helium for chromatography R.

Linear velocity: 50 cm/s.

Temperature:

|                | Time<br>(min)     | Temperature<br>(°C) |
|----------------|-------------------|---------------------|
| Column         | 0 - 14<br>14 - 54 | 80 → 220<br>220     |
| Injection port |                   | 250                 |
| Detector       |                   | 250                 |

Detection: flame ionisation.

Injection: 1  $\mu$ l.

Composition of the fatty acid fraction of the substance:

- myristic acid: maximum 5.0 per cent,
- palmitic acid: maximum 16.0 per cent,
- palmitoleic acid: maximum 8.0 per cent,
- stearic acid: maximum 6.0 per cent,
- oleic acid: minimum 58.0 per cent,
- linoleic acid: maximum 18.0 per cent,
- linolenic acid: maximum 4.0 per cent,

**Ethylene oxide and dioxan** (2.4.25, Method A): maximum 1 ppm of ethylene oxide and maximum 10 ppm of dioxan.

**Heavy metals** (2.4.8): maximum 10 ppm.

2.0 g complies with test C. Prepare the reference solution using 2 ml of lead standard solution (10 ppm Pb) R.

**Water** (2.5.12): maximum 3.0 per cent, determined on 1.00 g.

**Total ash** (2.4.16): maximum 0.25 per cent, determined on 2.0 g.

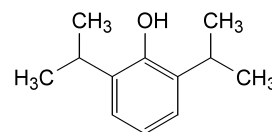
#### STORAGE

In an airtight container, protected from light.

04/2006:1558

## PROPOFOL

### Propofolum



$C_{12}H_{18}O$

$M_r$  178.3

#### DEFINITION

2,6-Bis(1-methylethyl)phenol.

Content: 98.0 per cent to 102.0 per cent.

This monograph applies to propofol prepared using distillation for purification.

#### CHARACTERS

Appearance: colourless or very light yellow, clear liquid.

Solubility: very slightly soluble in water, miscible with hexane and with methanol.

#### IDENTIFICATION

Infrared absorption spectrophotometry (2.2.24).

Comparison: propofol CRS.

#### TESTS

Refractive index (2.2.6): 1.5125 to 1.5145.

**Related substances.** Liquid chromatography (2.2.29).

Test solution (a). Dissolve 1.00 g of the substance to be examined in hexane R and dilute to 10.0 ml with the same solvent.

Test solution (b). Dissolve 0.240 g of the substance to be examined in hexane R and dilute to 100.0 ml with the same solvent.

Reference solution (a). Dissolve 5  $\mu$ l of the substance to be examined and 15  $\mu$ l of propofol impurity J CRS in hexane R and dilute to 50.0 ml with the same solvent.

Reference solution (b). Dilute 0.1 ml of propofol for peak identification CRS (containing impurities E and G) to 1.0 ml with hexane R.

Reference solution (c). Dilute 1.0 ml of test solution (a) to 100.0 ml with hexane R. Dilute 1.0 ml of this solution to 10.0 ml with hexane R.

Reference solution (d). Dissolve 0.240 g of propofol CRS in hexane R and dilute to 100.0 ml with the same solvent.

Column:

- size:  $l = 0.20$  m,  $\varnothing = 4.6$  mm;
- stationary phase: silica gel for chromatography R (5  $\mu$ m).

Mobile phase: anhydrous ethanol R, acetonitrile R, hexane R (1.0:7.5:99.0 V/V/V).

Flow rate: 2.0 ml/min.

Detection: spectrophotometer at 275 nm.

Injection: 10  $\mu$ l of test solution (a) and reference solutions (a), (b) and (c).

Run time: 7 times the retention time of propofol.